

Corrigendum

The Kinetics of Basic Cleavage of Nitrophenyl Alkanoate Esters by 'Hydroxypropyl- β -cyclodextrin' in Aqueous Solution

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In this paper, the working pH was given as 11.6. Judging from recent (and earlier) measurements on the same series of esters, it is clear that the pH was systematically in error, and closer to 11.5. Thus, the value of k_u , k_c and other rate constants are lower than they should be for pH 11.6. However, other quantities such as k_c/k_u , K_S and $K_{TS} = k_u K_S/k_c$ are not affected by pH and none of the trends or conclusions of the paper are altered.